

RELATIONSHIP OF COMPONENTS IN HOMOGENEOUS  
URANIUM MIXTURES

See page II.C.3-1 for a discussion of the general equation. Using this equation and the values in the table on page II.C.3-2 we can determine the particular equation:

U-H<sub>2</sub>O

$$H/U = \left( \frac{26089}{gU/l} - 1.3804 \right) / (1.00858f_{233} + f_{235} + .98736f_{238})$$

U Nitrate

$$H/U = \left( \frac{26089 - 600.2M}{gU/l} - 8.3467 \right) / (1.00858f_{233} + f_{235} + .98736f_{238})$$

(PuO<sub>2</sub>+UO<sub>2</sub>)-H<sub>2</sub>O

See page II.C.3-2